# SILICON-DIRECTED BECKMANN FRAGMENTATION

# HISAO NfSHIYAMA,\* KOjI SAWTA, NORtYUKI OSAKA, HfROYUKf ARAi, MAKOTO MATSUMOTO, and KENJf ITOH

School of Materials Science, Toyohashi University of Technology, Tempaku-cho, Toyohashi 440, Japan

*(Received in Japan 13 January 1988)* 

Abstract: The selective fragmentation reactions of *8-trimethylsilylketoximes* have been proved to proceed effectively with acid catalysts giving the corresponding nitriies. Cyclic sllylketoximes gave unsaturated nitriles. The fragmentation in the Beckmann rearrangement is completely controlled and directed by a trimethylsilyl group to lead the regio- and stereo-specific formation of the double bond. The catalytic fragmentation proceeds with the combination of trimethylsilyl ketoxime acetates and trimethylsilyl trifluoromethanesulfonate giving nitriles in high yields. Excellent stereospeclficitg of the fragmentation based on the stereochemical outcome was discussed. Simple stereo-controlled synthetic approach to some insect pheromones is also described.

The Beckmann reaction has been intensively studied and developed to a new synthetic methodology.<sup>1</sup> In numerous cases of the reaction, especially, the fragmentation has been observed for ketoximes having  $\alpha$ -substituents including hetero atoms, which can stabilize the intermediary carbocations.<sup>2</sup> The formation of nitriles is an important aspect for the fragmentatton. in particular, cyclic ketoxtmes are more attractive because of the fragmentation of w-functionalized nitriles. However, the control of the regioand stereo-chemistry of the forming double bonds has been difficult, because random deprotonation undergoes acidic media adopted.

In order to attain the regio- and stereo-specific formatlon of alkenyl function, it is necessary to recognize one specific proton out of others. For that purpose, trimethyisilyl group seems to be the most attractive candidate for a proton equivalent or a so-called "super proton"<sup>3</sup>. Recent interest in the utilization of the trialkylsilyl group as a leaving group lies in highly selective formation of target olefins.<sup>4</sup> In the present problem, if one can place a trimethylsilyl group at the B-carbon atom of the ketoximes. the Beckmann fragmentation may take place efficiently to give the desired double bond derived from the selective Si-C bond cleavage without the competing deprotonation. $<sup>5</sup>$ </sup>



## **2414 H.** NISHIYAMA et al.

# Fragmentation of (E)-ß-Trimethylsilylketoximes with Common Acid Catalysts.

The cyclic (E)- $\beta$ -trimethylsllylketoximes 1 and 2 gave the desired unsaturated nitriles 3 and 4, respectively, by the reactions with common acid catalysts for the Beckmann rearrangement in  $CH_2Cl_2$  at **room temperature for 1 day; catalyst** ( **yield of the olefin, %)**; for 1, PC1<sub>5</sub> (28%), P<sub>2</sub>O<sub>5</sub> (73%), POC1<sub>3</sub> (48%), MsCl-Pyridine (52%); for 2, P<sub>2</sub>O<sub>5</sub> (51%). The corresponding lactams, 5 and 6, were obtained in **535% yields. It is important to note that the Beckmann fragmentatton took place successfully under mild condition without formation of any other olefin isomers, since It has been well known that the frag-**



mentation via deprotonation is limited to the case of oximes having three alkyl groups or one aryl group on the  $a$ -carbon of the  $(E)$ -chain.<sup>6</sup> Recently, Hudrlik independently reported the effect of the tri-Recently, Hudrlik independently reported the effect of the tri**methylsiiyl group on the Beckmann rearrangement and fragmentation, which is closely related to our work.7** 

**We, therefore, examined the reaction of the non-sllylated oxime acetate 7, 2-methylcyclohexanone oxfme acetate, and the silylketoxime 8, respectively. Both of the oximes 7 and 8 were treated with**   $BF_{3}$ <sup> $\cdot$ </sup> Et<sub>2</sub>O. The oxime 7 gave only the lactam 9 (59%) at room temperature for 16 h. Whereas the **oxime 8** was converted selectively to the nitrile 3 (80%) at 0°C for 4 h. By the treatment with  $P_2O_5$ , **7** was recovered in 71% and the (Z)-oxime isomer was obtained in  $16\%$  in  $CH_2Cl_2$  at room temperature for 7 h. In contrast, the reaction of 8 with  $P_2O_5$  gave the nitrile 3 (66%) and the lactam 5 (21%) even at 0°C for 2 h.<sup>8</sup> The clear difference in the reaction was thus observed. Consequently, we think that the initial migration of the (E)-side chain should be accelerated and induced by the silyl group at the  $\beta$ position. It would also imply that the elimination of the leaving group (OAc) and subsequent alkyl



migration are enhanced by electron donating nature of the silyl group toward the back lobe  $\{\sigma^*\}$  of the **cleaving C-C bond.** 

This silicon-enhanced fragmentation can be modified when N-methyl-2-fluoropyridinium salt<sup>9</sup> was employed. Treatment of 1 with the pyridinium salt in  $CH_2Cl_2$  at  $0^{\circ}C$  for 1 h gave the desired nitrile 3 **In good to excellent yield (71%). Fluoride ion includlng the intermediary pyridinium salt 10 could effectively attack on the silyl group. Under the same condition the silylketoxime 2 gave only trans-**



5-hexenonitrile 4 in 64%. Stereoselective formation of the olefin was also observed in the mild condition.

Under the same reaction conditions employed for the (E)-Isomers above descirbed, the corresponding (Z)-isomers 11 and 12 did not react in CH<sub>2</sub>Cl<sub>2</sub> at 20°C with BF<sub>3</sub>. Et<sub>2</sub>O and P<sub>2</sub>O<sub>5</sub>. Although the acetate 13 did not react with BF<sub>3</sub>. Et<sub>2</sub>O at 0°C, treatment of 13 at refluxing temperature in CH<sub>2</sub>Cl<sub>2</sub> for 5 h gave the nitrile 14 in 95% yield. The fragmentation from the (Z)-oxime isomer was thus observed under more drastic conditions. In contrast, the (E)-isomer 15 induced the same fragmentation even at 0°C for only 20 min.



We examined the behavior of the linear oxime acetate  $16$  [a mixture of  $(E)$  and  $(Z)$ , 55:45]. When the mixture with BF<sub>3</sub>.Et<sub>2</sub>O in refluxing CH<sub>2</sub>Ci<sub>2</sub> for 2h,  $\beta$ -phenyipropionitrile (67%) and the amide 17 (21%) were obtained. Thus the fragmentation from the (Z)-isomer was also observed. However the linear oximes 16 and 19 gave 3-trimethylsilylpropanamldes 20 (75%) and 21 (74%), respecttvely, which are normal Beckmann products.<sup>10</sup> These observation can be accounted for the migratory aptitude of anti-alkyl or aryl groups being present to the acetoxy group. The facile anti-migration of a phenyl and a t-butyi group gave the corresponding amldes 20 and 21 with neither syn-migration or desilylative fragmentation. However, because of the low migratory aptitude of the phenethyl group in 16, upon heating (Z)-(E)-isomerization would occur prior to the migration and then the 2-trimethylsilylethyl group could migrate.<sup>11</sup>



**catalytic Design** 

We reasoned that as a special design of the silicon-directed fragmentation, the combination of the sllyiketoxime acetate and trimethylsilyl trifluoromethanesulfonate  $(TMSOTf)^{12}$  makes the reaction catalytic to afford the desired unsaturated nltriles. The reaction was carrted out with 10 mol% of TMSOTf in  $CH_2Cl_2$  at  $0^{\circ}C$  to room temperature for several hours. The obvious advantages of the results lead the specific formation of the nitriles in high yields (Table 1). The stereospecificity of the reaction was also clarified.



Table 1. Catalytic Fragmentation of Silylketoxime Acetates with TMSOTf.



Fluoride-Induced Reaction: Anionic Fragmentation?

The actd-catalyzed Beckmann fragmentation as mentioned above could be formally a cationic  $1,4$ heterolysis (A), or the reaction would actually be reduced to 1,2-heterolysis in stepwise mechanism (B). In contrast, we are interested in the possibility of an anionic-heterolysis induced by fluoride ion (C).



However, we could find no fragmentation of 1 and 8 by treatment with an excess of CsF in CH<sub>3</sub>CN at refluxing temperature. Therefore we have tried the two oxime isomers 37 and 38 having the silyl group at the benzylic position.<sup>13</sup> Treatment of the both isomers with an excess of CsF in CH<sub>3</sub>CN at refiuxing temperature gave only the trans-Isomer 46 as a fragmentation product (85% from 37 and 91% from 38, respectively) and 2benzylcyclohexanone oxime acetate 41 (8-10%) as a minor product. As we have found, use of TMSOTf gave stereospecifically the corresponding olefins, the cis-olefin from 37 and the trans-olefin from 38, respectively. Accordingly. the fragmentation with CsF can be accounted for the facile inversion of the transient carbanion species, **which were partly protonated** in the reaction media.



#### **Synthetic Application**

On the basis of the regio- and stereo-specificity of the silicon-directed Beckmann fragmentation, new synthetic approach for some insect pheromones is demonstrated. Synthesis of 46, a component of the sex pheromone of potatotuberworm moth (phthorimaea operculella)<sup>14</sup>, was started from the preparation of the trimethylsilyl ketone 42. Addition of 2-cyclopentenone to an excess (1.5 eq) of trimethylsilyllithium<sup>15</sup> in THF-HMPA at -78°C followed by treatment with 1.5 eq of tri-n-butyltin chloride and 1-bromo-2-octyne at -50°C to -40°C for 1 h to give the silylketone 42 in 67%. The trapping of the enolate intermediate as the stannyl enolate (or stannyl ketone) is thought to lead good to excellent yield for the a-alkylation in the sequencial vicinal double alkylation.<sup>16</sup> Hydrogenation of 42 with Pd-BaSO<sub>4</sub> and quinoline in methanol gave the cis-olefinic ketone 43 (84%). Oximation and subsequent acetylation of 43 gave the oxime acetate  $44$  (74%). Treatment of  $44$  with 10 mol% of TMSOTf at 0°C for 2.5 h afforded the nitrile 45 in 79% yield as a single product. Reduction of 45 with diisobutylaluminum hydride at -78°C and then with sodium borohydride at 0°C followed by acetylation gave the desired compound 46.

By utlliration of the nitriles 32 and 36 (in Table 1). a couple of the sex pheromones of Douglas fir tussock moth (orgyia psedotsugata) 17 could be synthesized. Alkylation **of** each unsaturated nitrile wlth n-decanyl Grignard reagent in ether followed by hydrolysis with hydrochloric acid gave the target ketones 47 (77%) and 48 (71%), respectively.

These sequences thus allow a unique approach to internal olefin synthesis with high stereospecificity via the stllcon-directed fragmentation.



# **Experimental Section.**

General: <sup>1</sup>H NMR spectra were recorded on JEOL JNM-FX90Q and JEOL JMR-PMX60 spectrometers using tetramethylsilane as the internal reference.  $^{13}$ C NMR spectra were obtained on a JEOL JNM-FXSOQ. High and low resolution mass spectra were determined with a JEOL JMS D-300 spectrometer. Infrared spectra were recorded on a JASCO A-3 spectrometer. GLC analyses were performed with a Shimadzu GC-3BT using a 2 m x 3 mm column (PEG 20M, 10%) and helium as carrier gas. Microanalyses were accomplished at the Microanalysis Center of Kyoto University. Analytical TLC was performed on Merck precoated silica gel plates (0.25 mm). Column chromatography was performed with silica gel (Merck Art 7734 and 9385).

All reactions were run under an inert atmosphere of Ar or  $N_2$ . Dichloromethane was dried by distillation under argon from CaCl<sub>2</sub> and phosphorus pentoxide. Ether and tetrahydrofuran were distilled from sodium and lithium alumlnum hydride. Other chemicals used were purchased.

# **Preparation of Silylketoxime 1 and its Acetate 8.**

**Alkylation of 2.5** mL (25.0 mmol) of cyclohexene oxide with trimethylsilylmethyl magnesium chloride (33.0 mmol) in 60 mL of THF in the presence of 350 mg of Cul at -20°C for 2 h gave 3.98 g of trans-2-trimethylsilylmethylcyclohexanol in 80% yield. The alcohol was purified by silica gel column chromatography and then was subjected to treatment with 4.5 g of CrO<sub>3</sub> in 9 mL of pyridine and 60 mL of  $CH_2Cl_2$  at room temperature for 1 h gave 2.19 g of 2-trimethylsilylmethylcyclohexanone in 69% yield. Treatment of 1.5 g (8.1 mmol) of the ketone with 7.0 g of HONH<sub>2</sub>·HCl and 1.4 g of NaOAc in 20 mL of ethanol at 0°C for lh gave 1.26 g (6.3 mmoll of (E)-P-trtmethylsilylmethylcyclohexanone oxime 1 (78 %) and 0.27 g (1.35 mmol) of the  $(Z)$ -isomer (17%); Rf, 0.55 for 1 and 0.40 for  $(Z)$ -isomer (hexane-ether = 1:1). 1: mp 42.0-43.0 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 90 MHz) -0.02 (s, 9H), 0.56 (dd, J = 8.1 and 15.0 Hz, 1 H, CH<sub>2</sub>Si), 0.96 (dd, J = 4.2 and 15.0 Hz, 1 H, CH<sub>2</sub>Si), 1.3-1.9 (m, 6 H), 2.3-2.6 (m, 3 H, H<sub>2</sub>C(6) and HC(2)) ;<sup>13</sup>C NMR (CDC1<sub>3</sub>, 22.4 MHz) -0.74 (q), 18.96 (t, CH<sub>2</sub>Si), 22.76 (t), 23.05 (t), 26.36(t), 35.66 (t), 38.19 (d.

HC(2) }, 163.94 (s, .C=N); IR (film) 3500-3000, 1640, 1238, 840 cm<sup>-1</sup>; Anal. Calcd for C<sub>10</sub>H<sub>21</sub>NOSi: C, 60.25 ; H, 10.62; N, 7.03. Found: C, 59.74; H, 10.94: N, 6.98. The  $(Z)$ -isomer 11: <sup>1</sup>H NMR (CDCl<sub>3</sub>, 90 MHz)  $-0.01$  (s, 9 H), 0.75 (dq, 2 H, CH<sub>2</sub>Si), 1.3-2.0 (m, 6 H), 2.1-2.3 (m, 2 H, H<sub>2</sub>C(5) ), 3.6 (m, 1 H. HC(2) ).

Treatment of 199 mg (1.0 mmol) of 1 with 0.15 mL of acetic anhydrlde and 0.5 mL of pyridlne at 0°C for 5 h gave 231 mg (0.96 mmol) of 8 (96%), which was purified by silica gel chromatography. Rf, 0.30 for 8 and 0.40 for 1 (hexane:ether = 1:4). 8: a colorless oil; <sup>1</sup>H NMR (CDC1<sub>3</sub>, 90 MHz) -0.02 (s, 9 H), 0.71 (dd, J = 8.5 and 14.8 Hz, 1 H,  $H_2$ CSI), 1.00 (dd, J = 6.8 and 14.8 Hz, 1 H,  $H_2$ CSI), 1.4-1.9 (m, 6 H), 2.12 (s, 3 H, CH<sub>3</sub>CO), 2.35-2.70 (m, 3 H); IR (film) 1755, 1636, 1245, 1200, 850 cm<sup>-1</sup>; Anal. Calcd for C<sub>12</sub>  $H_{23}$ NO<sub>2</sub>Si: C, 59.71; H, 9.60; N, 5.80. Found: C, 59.73; H, 9.60:, N, 5.92.

## Preparation of Silylketoxime 2 and its Acetate 30.

Starting trans-2-methyl-3-trimethylsilylcyclohexanone was prepared from 2-cyclohexenone by Still's procedure.<sup>15</sup> Treatment of 350 mg (1.90 mmol) of the silylketone with an excess of HONH<sub>2</sub>·HCl (3 mmol) and NaOAc (4 mmol) in EtOH as above mentioned gave 375 mg (1.88 mmol) of (E)-trans-2-methyl-3-(trimethylsilyl)cyclohexanone oxime 2 in 98% yiled. 2: mp 53.0-54.0 °C; <sup>1</sup>H NMR (CDC1<sub>3</sub>, 90 MHz) 0.0 (s, 9 H), 0.80-1.90 (m, 5 H), 1.13 (d, J = 7.0 Hz, <u>H<sub>2</sub>C(2)</u> ), 2.28 (dq, J<sub>HC(2)</sub>\_H<sub>C(3)</sub> = 8.5 Hz, J = 7.0 Hz, 1 H), 3.0 (m, 1 H, <u>H</u>C(6) ), 9.6 (b, 1 H); <sup>13</sup>C NMR (CDC1<sub>3</sub>, 22.5 MHz) -1.33 (q), 17.83 (q), 24.32 (t), 26.70 (t), 27.34 (t), 34.07 (d, C(3) ), 38.94 (d, C(2) ), 164.05 (s); IR (film) 3600-3000, 1660, 1250, 840 cm<sup>-1</sup>; Anal. Calcd for  $C_{10}H_{21}NOSi$ : C, 60.25; H, 10.62; N, 7.03. Found: C, 60.05; H, 10.26; N, 6.77.

Treatment of 260 mg (1.30 mmol) of 2 with 0.45 mL of acetic anhydride and 0.75 mL of pyrfdine at 0°C for 2 h gave 308 mg (1.28 mmol) of 30 tn 98% **yield. 30** was purified by silica gel column chromatography. 30: a colorless oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 90 MHz) 0.01 (s, 9 H), 0.80 (m, 1 H, HCS1), 1.22 (d, 3 H), 1.40-1.90 (m, 4 H), 2.13 (s, 3 H, CH<sub>3</sub>CO), 2.45 (dq, 1 H, HCCH<sub>3</sub>), 2.60-2.80 (m, 2 H, H<sub>2</sub>C(6) ); <sup>13</sup>C NMR  $(CDC1<sub>3</sub>, 22.5 MHz)$ , -1.43 (q), 18.51 (q, CH<sub>3</sub>), 20.02 (q, CH<sub>3</sub>CO), 25.53 (t), 26.87 (t), 27.85 (t), 33.63 (d, C (3) ), 38.69 (d, C(2) ), 168.90 (s, C=N), 172.14 (s, CO); IR (film) 1765, 1630, 1250, 840 cm<sup>-1</sup>; Anal. Calcd for **C12H23N02SI: C, 59.71; H, 9.80; N, 5.80.** Found: C, 60.03; H, **9.77; N, 8.07.** 

# **Preparation** of **Silylketoximes 12. 13, and 15.**

Starting 2-methylene-1-tetralone<sup>18</sup> was prepared by the method according to the literature. Transformation of the tetralone to the corresponding silylketone was carried out by Still's procedure. The silylketone (424 mg, 1.82 mmol) was treated with HONH<sub>2</sub><sup>+</sup> HCl (255 mg, 3.70 mmol) and NaOAc (430 mg, 5.20 mmol) in ethanol at room temperature for 40 h. Concentration of the reaction mixture under reduced pressure and extraction of the residue with ether gave a colorless oil, which was treated with 0.21 mL of acetic anhydride and  $0.74$  mL of pyridine in 1 mL of  $CH_2Cl_2$  at room temperature for 1.5 h. The reaction mlxture was concentrated, and the residual oil was purlfled by silica gel column to give 463 mg (1.60 mmol) of 13 and 15; Rf, 0.45 for 13 and 0.40 for 15 (ether:hexane = 1:4, two developments). 13: mp 46.0-46.5 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 90 MHz) 0.0 (s, 9 H), 0.71 (m,  $H_2$ CSi), 1.55-2.00 (m, 2 H, HC-C=N), 2.18 (s, 3 H,  $H_3$ CC=O), 2.70 (t , 1 H), 2.88 (t, 1 H), 7.0-7.35 (m, 3 H, aromatic), 8.00 (dd, 1 H, aromatic); <sup>13</sup>C NMR  $(CDC_{13}^7, 22.5 \text{ MHz})$  -0.98, 17.7, 20.8, 24.4, 27.8, 29.4, 126.4 (two carbons), 128.7, 130.5, 139.4, 166.3, 168.8; IR (film) 1770, 1362, 1250 cm<sup>-1</sup>; Anal. Calcd for C<sub>16</sub>H<sub>23</sub>NO<sub>2</sub>Si: C, 66.39; H, 8.01; N, 4.84. Found: C, 66.17; H, 7.91; N, 4.78.

# **Preparation** of Sllylketoxime Acetate 33.

According to Still's procedure, the starting cis-2-methyl-3-(trlmethylsilyl)cyclohexanone was prepared from 2-methyl-2-cyclohexenone<sup>19</sup>. To a solution of 2.2 mL (10.5 mmol) of hexamethyldisilane and 9 mL of HMPA at 0°C was added a ether solution of methyllithium (9.5 mmol). The mixture was stirred for 20 min. dilutad 10 mL of THF, and then was cooled to -78'C. A solution of 2-methyl-2-cyclohexenone (098 mg, 6.34 mmol) In 2 mL of THF was added. Tbe mixture was stirred for 30 min and 5 mL of water was added. The mixture was extracted with a mixture of ether and hexane (1:l). The organic layer was washed with water, dried over anhydrous  $MgSO<sub>A</sub>$ , and concentrated. The residual oil was purified by silica gel column to give 644 mg (3.50 mmol) of cis-2-methyl-3-(trimethylsilyl)cyclohexanone as colorless oil: IR

(film) 1700, 1250 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 60 MHz) 0.11 (s, 9 H), 1.23 (d, 3 H), 1.53-2.17 (m, 5 H), 2.27-2.80 (m, 3 H). The ketone (447 mg, 2.43 mmol) was treated with HONH<sub>2</sub><sup>+</sup> HCl (290 mg, 4.2 mmol) and NaOAc (570 mg, 7.0 mmol) in 15 mL of ethanol at 0°C for 1.5 h to give the (E)-oxime (151 mg, 0.76 **mmol) and the (Z)-oxime (163 mg, 0.82 mmol); & 0.42 for the (E)-oxime and 0.31 for the (Z)-oxime**  (ether:hexane = 1:3, two developments). The (E)-oxime: a colorless oil; <sup>1</sup>H NMR (CDC1<sub>3</sub>, 90 MHz) 0.0 (s, **9 H), 1.10 (d, 3 H, J = 7.5 Hz,**  $H_3C$ **), 1.30-2.10 (m, 5 H), 2.50-2.85 (m, 2 H,**  $H_2C(2)$ **,**  $H_2C(6)$  **), 3.18 (broad, 1 H,**  $\underline{H}C(6)$  **); IR (film) 3300, 1660, 1375, 1250 cm<sup>-1</sup>; Anal. Calcd for C<sub>10</sub>H<sub>21</sub>NOSi: C, 60.22; H, 10.62; N, 7.03. Found: C, 60.22; H, 10.82; N, 6.81. 10 ;Ll The (Z)-oxime: a colorless oil; H NMR (CDCI,, 90 MHz), 0.0**  (s, 9 H), 1.05 (d, 3 H,  $H_3C$ ), 1.40-2.10 (m, 5 H), 2.10-2.35 (m, 2 H,  $H_2C(6)$ ), 3.70 (dq, 1 H, J = 3.0 and 7.5 Hz,  $\underline{H}C(2)$  ); IR (film) 3300, 1660, 1375, 1250 cm<sup>-1</sup>.

**The (E)-oxime (145 mg, 0.73 mmol) was treated with 0.1 mL of acetic anhydride and 0.18 mL of**  pyridine in  $CH_2Cl_2$  (0.5 mL) at 0°C for 1 h to give 143 mg (0.59 mmol) of the acetate 33 (81%) as a **colorless oil. 2 T 33: H NMR (CDC13, 60 MHz) 0.05 (s, 9 H), 1.20 (d, 3 H, fi,C), l-40-2.40 (m, 6 H), 2.18 (s, 3 H, H3CCO), 2.93-3.40 (m, 3 H); 13C NMR (CDC13, 22.5 MHz) -2.0. 16.0, 19.9, 21.3, 23.0, 28.7, 32.2,**  36.6, 165.7, 172.9; IR (film) 1765, 1650, 1250 cm<sup>-1</sup>; Anal. Calcd for C<sub>12</sub>H<sub>23</sub>NO<sub>2</sub>Si: C, 59.71; H, 9.60. **Found: C, 59.26; H, 9.80.** 

## **Preparation of Silylkeoxime Acetate 28.**

Starting 1-benzyl-2-(trimethylsilyl)-2-cyclopentanone, as described above, was prepared with 2-cyclo**pentenone, benzyl bromide, and trimethylsilyllithlum. The ketone (310 mg, 1.26 mmol) was treated with HONH2,HCI (104 mg, 1.50 mmol) and NaOAc (204 mg, 2.50 mmol) in 3 mL of ethanol at 0°C for** 1 **h to give 257 mg (1.0 mmol) of the (El-oxime and 64 mg (0.25 mmol) of the (Zf-oxlme; the (El-oxlme: mp 68.0 -70.0 "C. The (E)-oxime (210 mg, 0.80 mmol) was treated with 0.1 mL of acetic anhydride and 0.18 mL of pyridine at 0°C for 3 h to glve 218 mg (0.72 mmol) of the acetate 28. 28: a colorless oil: 'H NMR**  (CDCI<sub>3</sub>, 90 MHz) 0.02 (s, 9 H), 1.20 (m, 1 H, HCSi), 1.50-2.80 (m, 5 H), 2.20 (s, 3 H, H<sub>3</sub>C), 3.05 (s, 2 H,  $H_2$ CPh), 7.30 (s, 5 H); IR (film) 1765, 1655, 1250, 835, 750, 700 cm<sup>-1</sup>; Anal. Calcd for C<sub>17</sub>H<sub>25</sub>NO<sub>2</sub>Si: C, **67.28; H, 8.30; N, 4.62. Found: C, 66.98; H, 8.16; N, 4.71.** 

# **Preparation of Sllylketoxime Acetate 16.**

**Treatment of 4-phenyl-I-butene oxide with trimethylsilyimethyl Grignard reagent in ether in the presence of CuI (10 mol%) gave 5-phenyl-1-trimethylsilyl-3-pentanol in 95% yield as a** *colorless* **oil.**  Oxidation of the alcohol (0.67 g) with CrO<sub>3</sub> (1.7 g) and pyridine (3.2 mL) in CH<sub>2</sub>Cl<sub>2</sub> (25 mL) for 1.5 h gave 0.50 g of 5-phenyl-1-trimethylsilyl-3-pentanone in 74% yield. The ketone (0.48 g, 2.07 mmol) was **treated with HONH2. HCL (0.34 g) and NaOAc (1.1 g) in 10 mL of ethanol at (PC for** I **h gave the oxime as a colorless oil. The crude product was directiy acetylated. Treatment of the oxime with 0.6 mL of acetic anhydride and 0.8 mL of pyridlne at room temperature for 1 day gave the silylketoxime acetate 16 (0.50 g, 1.72 mmol); a mixture of (E) and (Z)-isomers.** The mixture: colorless oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>, **90 MHz) 0.0 (s), 0.03 (s), 0.43-0.99 (m, 2 H,**  $\frac{11}{2}$ **CSi), 2.08 (s, 3 H,**  $\frac{11}{3}$ **CCO), 2.53-3.11 (m, 4 H, PhC** $\frac{11}{2}$ **CH<sub>2</sub>),** 7.31 (s, 5 H); IR (film) 1760, 1630, 1250, 840 cm<sup>-1</sup>; Anal. Calcd for C<sub>16</sub>H<sub>05</sub>NO<sub>2</sub>Si: C, 65.93; H, 8.65; N, **4.81. Found: C, 66.16; H, 8.53; N, 4.66.** 

#### **Preparation of Silylketoxime Acetate 18.**

**To a solution of 945 mg (7.0 mmol) of acetophenone oxime in THF (15 mL) at O'C was added**  n-butyHithium(1.5 N-hexane, 9.8 mL). The mixture was stirred for 1 h at 0°C. A solution of 1.56 g (7.3 **mmol) of iodomethyltrimethylsilane in 3 mL of THF was added and the mixture was stirred for 2 h. After usual work-up, 522 mg of phenyl 2-trimethylsilylethyl ketoxfme was obtained. Treatment of the oxime (278 mg, 1.26 mmol) with 0.24 mL of acetic anhydride and 0.61 mL of pyridine at 0°C for 2 h gave the silylketoxime acetate 18 (0.292 mg, 1.03 mmol). 18: mp 69.5-70.5 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 60 MHz) 0.04** 

(s, 9 H), 0.50-0.96 (m, 2 H,  $H_2CSi$ ), 2.25 (s, 3 H,  $H_3CCO$ ), 2.60-2.98 (m, 2 H,  $H_2CCN$ ), 7.50 (m, 5 H); **IR** (film) 1770, 1613, 1575, 1250, 860, 840 cm<sup>-1</sup>; Anal. Calcd for  $C_{14}H_{21}NO_2Si$ : C, 63.84; H, 8.04; N, **5.32.** Found: C, 63.83; H, 7.93; N, 5.40.

# Silicon-directed Beckmann fragmentation 2421

#### **Preparation of Siiylketoxime Acetate 19.**

**AS** described for **18,** dlllthiate of plnacolone oxlme gave r-butyl a-trimethylsllylethyl ketoxlme In a low yield. The oxime (347 mg, 1.72 mmol) was treated wlth 0.33 mL of acetic anhydride and 0.83 mL of pyridine at room temperature for 2 h gave the silylketoxime acetate **19** (417 mg, 1.72 mmol) as a COlOrless oil. 19: <sup>1</sup>H NMR (CDC1<sub>2</sub>, 60 MHz) 0.10 (s, 9 H), 0.60-0.99 (m, 2 H,  $H_2$ CSi), 1.33 (s, 9 H), 2.18 (s, 3 H,  $H_3$ CCO), 2.20-2.43 (m, 2 H,  $H_2$ CCN); IR (film) 1772, 1621, 1370, 1253, 865, 833 cm<sup>-1</sup>; Anal. Calcd for  $C_{12}H_{25}NO_2Si: C, 59.21; H, 10.35; N, 5.75.$  Found: C, 59.41; H, 10.23; N, 5.87.

## **Preparation of Silylketoxime Acetate 22.**

To a solution of 1 (199 mg, 1.0 mmol) in 3 mL of THF was added n-butyllithium  $(1.5 \text{ N}$  in hexane, **1.4 mL). 'i'he** mixture was stirred at room temperature for 1 h and lodomethane (260 me) was added. After stirring for 1 h and usual work-up, cis-6-methyl oxime (191 mg, 0.90 mmol) was obtained; Anal. Calcd for C<sub>11</sub>H<sub>23</sub>NOSi: C, 61.91; H, 10.86; N, 6.56. Found: C, 61.90; H, 11.17; N, 6.34. The oxime (227 mg, 1.07 mmol) was treated with 0.3 mL of acetic anhydride and 1.0 mL of pyridine at 0°C for 1 h to give 215 mg (0.84 mmol) of 22 as a colorless oil. 22:  $^{1}$ H NMR (CDCl<sub>3</sub>, 90 MHz) -0.02 (s, 9 H), 0.45 (m, 1 H), 0.62 (m, 1 H), 1.10 (d, 3 H, <u>H</u><sub>3</sub>C), 1.20-1.80 (m, 6 H), 2.14 (s, 3 H, <u>H<sub>3</sub>CCO), 2.55 (m, 1 H, H</u>CCSi), 3.55 (m,1 H, <u>H</u>CCH<sub>3</sub>); IR (film) 1763, 1618, 1250, 840 cm <sup>-</sup>; MS, <u>m/e</u> 255 (2, M), 240 (50), 198 (100), 180 (90), 152 (35), 140 (30).

### Preparation of Silylketoxime Acetate 24.

To a solution of trimethylsilylmethyl magnesium chloride (9.0 mmol) in 20 mL of THF was added norbornene oxtde (584 mg, 5.31 mmol) and 257 mg of CuI. The mixture **was** stirred at **room temperature**  for 16 h. After usual work-up, silylalcohol (458 mg, 2.31 mmol) was obtained. Subsequent oxidation of the alcohol (452 mg, 2.30 mmol) with CrO<sub>3</sub> (460 mg) and pyridine (0.74 mL) in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) at room temperature for 9 h to give the corresponding ketone (400 mg, 2.04 mmol) as a colorless 011; IR (film) 1740 cm<sup>-1</sup>. The ketone (389 mg, 1.98 mmol) was treated with HONH<sub>2</sub> HCl (165 mg) and NaOAc (325 mg) **in** 5 mt of ethanol to give the oxlme (368 mg, 1.74 mmol). The oxime (266 mg, 1.26 mmol) was treated wlth 0.14 mL of acetic anhydrlde and 0.8 mL of pyrfdine to give the silyiketoxime acetate 24 (304 **mg,**  1.20 mmol) as a colorless oil. 24: <sup>1</sup>H NMR (CDC1<sub>3</sub>, 60 MHz) 0.0 (s, 9 H), 0.63 (dd, J = 10.0 and 15.0 Hz, 1 H, HCSi), 1.06 (dd, J = 5.0 and 15.0 Hz, 1 H, HCSi), 1.20-1.50 (m, 6 H), 2.10 (s, 3 H, H<sub>3</sub>CCO), 2.10-2.30 (m, 1 H), 2.55 (m, 1 H, HCCSi), 3.35 (m, 1 H, HCCN); IR (film) 1765, 1620, 1250, 840 cm<sup>-1</sup>; Anal. Calcd for C<sub>13</sub>H<sub>23</sub>NO<sub>2</sub>Si: C, 61.62; H, 9.15; N, 5.53. Found: C, 61.37; H, 9.24; N, 5.52.

# Preparation of Silylketoxime Acetate 26.

Alkylation of 1.0 g of 2-carboethoxycyclododecanone with NaH (1.2 eq) and iodomethyl(trimethyl)silane  $(2.0 \text{ eq})$  in 6 mL of benzene and 3 mL of DMF at 60°C for 2 days gave 390 mg of 2-carboethoxy-2-(trimethylsilylmethyl)cyclododecanone. Treatment of 333 mg of the ketone with excess Ba(OH)<sub>2</sub> in methanol at reflux temperature for 1 week gave 100 mg (0.30 mmol) of 2-(trimethylsilylmethyl)cyclododecanone. The silylketone (61 mg, 0.23 mmol) was treated with HONH<sub>2</sub> HCI (20 mg) and NaOAc (40 mg) in 1 mL of ethanol at 80°C for 1 h to give the oxime (60 mg). The crude oxime was treated with 0.05 mL of acetic anhydride and 0.1 mL of pyridine in  $CH_2Cl_2$  (1 mL) to give the silylketoxime acetate 26 (59.5 mg, 0.18 mmol). 26: <sup>1</sup>H NMR (CDCI<sub>3</sub>, 60 MHz) 0.0 (s, 9 H), 0.90 (m, 2 H,  $H_2$ CSi), 1.0-2.0 (m, 18 H), 2.10 (s, 3 H,  $H_3$ CCO), 2.20-2.80 (m, 3 H); IR (film) 1765, 1620, 1250, 840 cm<sup>-1.</sup>

# Preparation of Silylketoxime Acetates 37 and 38.

To a solution of 2.30 g (14.0 mmol) of benzyltrimethylsilane and N,N,N',N'-tetramethylethylenedlamine (3mL) in 30 mL of THF at -78°C was added n-butyllithium (in hexane, 1.68 N, 8.33 mL). After stirring at 0°C for 30 min, the mixture was then cooled at -78°C. Cyclohexene oxide (0.98 g, 10.0 mmol) was added to the mixture. The temperature was then raised to -15°C. After stirring for 4 h, aqueous ammonium chloride was added and the mixture was extracted with ether. The organic layer was dried over anhydrous magnesium sulfate and concentrated to give the residual oil, which was purified by silica

gel column chromatography. 2-[(Trimethylsilyl)phenylmethyl]-cyclohexanol (2.03 g, 7.72 mmoi) was obtained In 77% yield. The alcohol was treated with CrO<sub>3</sub> (2.0 g) and pyridine (5.0 mL) in 30 mL of CH<sub>2</sub>Cl<sub>2</sub>. The mixture was stirred for 4 h. Ether (50 mL) was added to the mixture. The extract was concentrated and purified by silica gel to give the two isomers of cyclohexanone derivatives (threo ketone, 795 mg; erythro ketone, 785 mg; total 82%; Rf, 0.63 and 0.56, respectively). Threo ketone: a colorless oil; <sup>1</sup>H NMR (CDCI<sub>q</sub>, 90 MHz) -0.10 (s, 9 H), 1.01-2.11 (m, 6 H), 2.27-2.68 (m, 2 H,  $\underline{H}_2$ CCO), 2.52 (d, 1 H, J  $= 10.6$  Hz, HCSi), 2.75-3.10 (m, 1 H, HCCO), 6.81-7.37 (m, 5 H); IR (film) 1712, 1599, 1496, 1246, 738, 698 cm<sup>-1</sup>; Anal. Calcd for C<sub>16</sub>H<sub>24</sub>OSi: C, 73.79; H, 0.29. Found: C, 73.93; H, 9.35. Erythro ketone: mp **38.0-39.0 T;** IH NMR (CDC13, 90 MHz) -0.03 (s. 9 HI. 1.47-2.44 (m, 8 HI, 2.55 (d, I H, J = 9.2 Hz, EC-Si), 2.69-3.18 (m, 1 H, HCCO), 6.85-7.33 (m, 5 H); Anal. Calcd for  $C_{16}H_{24}$ OSi: C, 73.79; H, 9.29. Found: C, 73.52; **H, 9.36.** As above described, the two isomers of the ketones were transformed to the oximes, which were subsequently acetylated: threo ketone (615 mg, 2.36 mmol), HONH<sub>2</sub>. HCI (197 mg), NaOAc (387 mg), EtOH (8.0 mL), at room temp., for 3 h, threo oxime (623 mg, 2.26 mmol, 96%); erythro ketone (604 mg, 2.32 mmol), HONH<sub>2</sub>·HCl (193 mg), NaOAc (381 mg), EtOH (8.0 mL), at room temp., for 3 h, erythro oxime (611 mg, 2.22 mmol0, 96%); threo oxime (300 mg, 1.09 mmol), acetic anhydride (0.22 mL), pyridine (0.43 mL),  $CH_2Cl_2$  (2.0 mL), at 0°C, for 2 h, threo silylketoxime acetate 37 (335 mg, 1.06 mmol, 97%); erythro oxime (299 mg, 1.08 mmol), acetic anhydride (0.22 mL), pyridine (0.42 mL), CH<sub>2</sub>Cl<sub>2</sub> (2.0 mL), at 0°C, for 2 h, erythro silylketoxime acetate 38 (337 mg, 1.06 mmol, 98%); Rf, 0.56 for 37 and 0.48 for 38, ether:hexane = 1:1. 37: mp 76.0-77.0 °C; <sup>1</sup>H NMR (CDCI<sub>3</sub>, 90 MHz) -0.09 (s, 9 H), 1.06-2.31 (m, 7 H), 2.20 (s, 3 H,  $H_3$ CCO), 2.54 (d, 1 H, J = 12.5 Hz, HCSI), 2.93-3.41 (m, 2 H,  $H_2$ CCN), 6.80-7.41 (m, 5 H); IR (film) 1770, 1636, 1601, 1494, 1249, 741, 701 cm<sup>-1</sup>; Anal. Calcd for C<sub>18</sub>H<sub>27</sub>NO<sub>2</sub>Si: C, 68.09; H, 8.57; N, 4.41. Found: C, 68.87; H, 8.74; N, 4.36. 38: mp 75.0-76.0 °C; <sup>1</sup>H NMR (CDCI<sub>3</sub>, 90 MHz) -0.03 (s, 9 H), 1.08-2.30 (m, 7 H), 1.98 (s, 3 H, H<sub>3</sub>CCO), 2.58 (d, 1 H, J = 11.9 Hz, HCSi), 2.71  $(m, 1 H)$ , 3.25  $(m, 1 H)$ , 6.87-7.30  $(m, 5 H)$ ; IR (film) 1762, 1637, 1598, 1491, 1247, 720, 699 cm<sup>-1</sup>; Anal. Calcd for  $C_{18}H_{27}NO_2St$ : C, 68.09; H, 8.57; N, 4.41. Found: C, 67.87; H, 8.74; N, 4.36.

# Reaction of Silyiketoxime 1 with  $P_2O_5$ .

To a solution of 99 mg (0.50 mmol) of 1 in  $CH_2Cl_2$  (2.0 mL) was added ca. 70 mg (0.5 mmol) of phosphorus pentoxide at 0°C. The mixture was stirred for 1 day at room temperature. Ether (2 mL) and triethylamine (0.2 mL) were added and the mixture was washed with brine and concentrated under reduced pressure by aspirator. The residual oil was diluted with  $CDCI<sub>3</sub>$  (0.3 mL) and benzene (0.030 mL) as an Internal standard was added. Crude yield was calculated on the basis of integral value of olefinic protons by  ${}^{1}$ H NMR analysis (82%). The crude oil was then passed through a short column of silica gel to give 43 mg (0.37 mmol) of 3 (73%). Further purification for Identification was performed by preparative GLPC. 3: <sup>1</sup>H NMR (CDC1<sub>3</sub>, 60 MHz) 1.20-2.80 (m, 8 H), 4.90-5.20 (m, 2 H), 5.85 (m, 1 H); <sup>13</sup>C NMR  $(CDCI_{31}$ , 22.5 MHz) 17.17, 24.98, 27.97, 32.95, 115.52, 137.68; IR (film) 2250, 910 cm<sup>-1</sup>; MS exact mass  $m/e$  109.0873 (calcd for C<sub>7</sub>H<sub>11</sub>N, 109.0890); MS, m/e 109 (9, M), 73 (16).

# Reaction of Silylketoxime 2 with  $P_2O_5$ .

The silylketoxime 2 (124 mg,  $0.62$  mmol) was treated, as described for I, with phosphorus pentoxide (90 mg) in CH<sub>2</sub>Cl<sub>2</sub> (2.0 mL) at room temperature for 4 h to give 4 (35 mg, 0.32 mmol) in 51%. 4: <sup>I</sup>H NMR (CDCI<sub>3</sub>, 90 MHz) 1.50 (d, 3 H), 1.60 (m, 2 H), 2.0 (m, 4 H), 2.18 (t, 3 H, H<sub>2</sub>CCN), 5.50 (m, 2 H); <sup>13</sup>C NMR (CDC1<sub>3</sub>, 22.5 MHz) 16.45 (t, CCN), 17.96 (q), 127.47, 128.69 ; IR (film) 2250, 965 cm<sup>-1</sup>; MS exact mass m/e 109.0858 icalcd for C<sub>7</sub>H<sub>11</sub>N, 109.0890); MS, m/e 109 (100, M), 83 (11), 81 (58), 69 (41), 67(54).

# Reaction of Silylketoxime Acetate 8 with  $BF_3 \cdot Et_2O$ .

The acetate 8 (136 mg, 0.57 mmol) was treated with borontrifluoride etherate (0.077 mL) in CH<sub>2</sub>Cl<sub>2</sub>  $(1 \text{ mL})$  at  $0^{\circ}$ C for 4 h to give 3 (50 mg, 0.46 mmol) in 80%.

#### Reaction of 1 with N-Methyl-2-fluoropyridinium Iodide.

To a solution of 199 mg (1.0 mmol) of 1 in  $CH_2Cl_2$  (3.0 mL) was added 0.11 (1.1 mmol) of triethylamine and the pyridinium salt (263 mg, 1.10 mmol) at  $0^{\circ}$ C. The mixture was stirred for 1 h. After work-up as described above, 77 mg (0.71 mmol) of 3 was obtalned.

# Reaction of Silylketoxime Acetate 13 with BF<sub>3</sub>·Et<sub>2</sub>O.

The acetate 13 (132 mg, 0.35 mmol) was treated with borontrifluoride etherate (0.052 mL) in CH<sub>2</sub>Cl<sub>2</sub> (2.0 mL) at refluxing temperature for 5 h. After extraction with ether, the crude 011 was purified by silica gel column to give 56 mg (0.34 mmol) of the nitrile 14 as a colorless oil (97%). 14:  $^{1}$ H NMR (CDCl<sub>3</sub>, 60 MHz) 2.34 (m, 2 H), 3.48 (m, 2 H), 4.70-5.24 (m, 2 H), 5.85 (m, 1 H), 7.0-7.7 (m, 4 H); IR (film) 2200, 1600,1585, 905, 758 cm<sup>-1</sup>; MS <u>m/e</u> 158 (M), 117, 90, 40.

# Reaction of Silylketoxime Acetate 18 with BF<sub>3</sub> · Et<sub>2</sub>O.

The acetate 18 (77.4 mg, 0.29 mmol) was treated with borontrifluoride etherate (0.044 mL) in CH<sub>2</sub>Cl<sub>2</sub> (1.0 mL) at room temperature for 8 h. N-Phenyi-3-trimethylsilylpropanamide 20 (49 mg, 0.22 mmol) was obtained. 20: a colorless oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 60 MHz) 0.04 (s, 9 H), 1.00 (m, 2 H,  $\frac{H_2}{S}$ CSi), 2.39 (m, 2 H,  $H_2$ CCO), 7.10-7.77 (m, 5 H); IR (film) 1660, 1600, 1550, 1500, 1255, 860, 833 cm<sup>-1</sup>.

# General Procedure of Catalytic Fragmentation.

To a solution of the silylketoxime acetate (0.5-1.0 mmol) in anhydrous CH<sub>2</sub>Ci<sub>2</sub> (2-4 mL) was added dropwise TMSOTf (10 mol% equivalent) at O°C. The mixture was stirred for l-4 h (monitored by TLC). was treated with aqueous sodium bicarbonate and triethylamine (0.1 mL), and was quickly extracted with ether (10 mL). The extract was concentrated, and the residual oil was purified by silica gel column chromatography to give the corresponding nitrile.

## Reaction **of Sllylketoxime Acetate 8 with TMSOTf.**

The acetate 8 (121 mg, 0.50 mmol) was treated with TMSOTf (ca. 0.01 mL) in CH<sub>2</sub>Cl<sub>2</sub> (2.0 mL) at 0°C for 4 h. After work-up, the crude oil was diluted in CDC1 $_3$  and benzene (0.020 mL). Calculation on the basis of  $<sup>1</sup>H$  NMR analysis gave 89% yield of 3.</sup>

## **Reaction of Silylketoxlme Acetate 24 with** TWOTf.

The acetate 24 (134 mg, 0.53 mmol) was treated with TMSOTf  $(0.01 \text{ mL})$  in CH<sub>2</sub>Cl<sub>2</sub> (2.0 mL) at 0°C for 1 h. After work-up and silica gel chromatography, the nitrile 25 (60.9 mg, 0.50 mmol) was obtained. 25: a colorless oil; <sup>1</sup>H NMR (CDCI<sub>q</sub>, 90 MHz) 1.40-2.10 (m, 6 H), 2.40 (m, 1 H, HCC=C), 2.82 (m, 1 H,  $\text{HCC=N}$ , 4.90-5.20 (m, 2 H), 5.80 (m, 1 H); IR (film) 2230, 1640, 990, 910 cm<sup>-1</sup>; MS exact mass m/e 121.0889 (calcd for  $C_8H_{11}N$ , 121.0890).

The reaction of other silylketoxime acetate 22, 26, 28, 30, 31, 33, and 34 with TMSOTf were carried out in a manner similar to that described above. The reaction conditions and analytical data of the nitriles are listed below.

22 (122 mg, 0.48 mmol), TMSCTf (0.01 mL), CH<sub>2</sub>Cl<sub>2</sub> (2 .0 mL), 0°C, 2 h; 23 (48 mg, 0.39 mmol): a colorless oil; <sup>1</sup>H NMR (CDC1<sub>3</sub>, 60 MHz) 1.30 (d, 3 H), 1.50-1.80 (m, 4 H), 2.10 (m, 2 H), 2.60 (m, 1 H), 4.80-5.20 (m, 2 H), 5.85 (m, 1 H); IR (film) 2220, 1640, 940, 910 cm<sup>-1</sup>; MS exact mass  $m/e$  123.1084 (calcd for  $C_8H_{13}N$ , 123.1048).

26 (52 mg, 0.16 mmol), TMSOTf (0.005 mL), CH<sub>2</sub>Cl<sub>2</sub> (0.8 mL), 0°C, 2 h; 27 (25 mg, 0.128 mmol): a colorless oil; <sup>1</sup>H NMR (CDC1<sub>3</sub>, 60 MHz) 1.20-2.80 (m, 20 H), 4.80-5.20 (m, 2 H), 5.80 (m, 1 H); IR (film) 2250, 1640, 990, 910 cm<sup>-1</sup>; MS exact mass m/e 193.1837 (calcd for C<sub>13</sub>H<sub>23</sub>N, 193.1831).

28 (147 mg, 0.48 mmol), TMSOTf (0.01 mL), CH<sub>2</sub>Cl<sub>2</sub>(2.0 mL), 0°C, 3 h; 29 (76.3 mg, 0.45 mmol): a colorless oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 60 MHz) 2.43 (broad s, 4 H), 3.44 (d, 2 H), 5.17-5.95 (m, 2 H), 7.21 (s, 5 H); IR (film) 2250, 964 cm<sup>-1</sup>; MS exact mass m/e 171.1042 (calcd for  $C_{12}H_{12}N$ , 171.1037)

30 (121 mg, 0.50 mmol), TMSOTf (0.01 mL), CH<sub>2</sub>Cl<sub>2</sub> (2.0 mL), 0°C, 3 h; 4 (51 mg, 0.47 mmol). 31 (93 mg, 0.31 mmol), TMSOTf (0.006 mL),  $CH_2^CCl_2^T$  (1.0 mL), 0°C, 1 h; 32 (47.2 mg, 0.29 mmol): <sup>1</sup>H NMR (CDCl<sub>3</sub>, 90 MHz) 0.89 (t, 3 H), 1.10-1.50 (m, 8 H), 1.70 (m, 2 H), 2.05 (m, 2 H), 2.32 (t, 2 H, H<sub>2</sub>C CN), 5.10-5.70 (m, 2 H); <sup>13</sup>C NMR (CDCI<sub>3</sub>, 22.5 MHz) 14.1, 16.3, 22.6, 25.3, 20.2, 31.3, 32.6, 119.0, 127.3, 133.3; IR (film) 2245, 965 cm<sup>-1</sup>; MS m/e 166 (20, M+1), 165 (14, M), 137 (56), 136 (65), 122 (70), 108 (33), 95 (34), 80 (100), 67 (53), 55 (80); MS exact mass m/e 165.1516 (calcd for  $C_{11}H_{10}N$ , 165.1516). 33 (128 mg, 0.53 mmol), TMSOTf (0.01 mL), CH<sub>2</sub>Cl<sub>2</sub> (2.0 mL), 0°C, 3 h; 35 (54.0 mg, 0.50 mmol): <sup>1</sup>H NMR (CDCI<sub>3</sub>, 90 MHz) 1.50-1.90 (m,4 H), 2.35 (m, 2 H, H<sub>2</sub>CC=C), 2.40 (t, 2 H, H<sub>2</sub>CCN), 5.40 (m, 2 H); <sup>13</sup>C NMR (CDCI<sub>3</sub>, 22.5 MHz) 12.86 (q, CH<sub>3</sub>, cis-methyl), 16.46 (t), 25.31 (t), 25.65 (t), 126.36, 128.04; IR (film) 2240, 1655, 910, 835 cm<sup>-1</sup>; MS exact mass m/e 109.0898 (calcd for C<sub>17</sub>H<sub>11</sub>N, 109.0890). 34 (70.6 mg, 0.24 mmol), TMSOTf (0.005 mL), CH<sub>2</sub>Cl<sub>2</sub> (1.0 mL), 0°C, 1 h; 36 (35.4 mg, 0.21 mmol): a colorless oil; <sup>1</sup>H NMR (CDC1<sub>3</sub>, 90 MHz) 0.89 (t, 3 H), 1.10-1.95 (m, 8 H), 1.70 (m, 2 H), 2.10 (m, 2 H), 2.34 (t, 2 H), 5.10-5.70 (m, 2 H);  $^{13}$ C NMR (CDCl<sub>3</sub>, 22.5 MHz) 14.1, 16.5, 22.6, 25.5, 26.0, 27.2, 29.4, 31.5, 119.8, 126.7, 132.6; MS <u>m/e</u> 166 (8, M+1), 165 (10, M), 137 (64), 134 (75), 122 (66), 95 (64), 80 (100), 67 (80), 55 (100); MS exact mass m/e 165.1499 (calcd for C<sub>11</sub>H<sub>19</sub>N, 165.1516).

#### **Reaction of Siiyiketoxlme Acetate 37 and 38 with CsF and TMSOTf.**

The acetate 37 (95.2 mg, 0.30 mmol) was treated with CsF (137 mg, 0.90 mmol) In acetonitrile (1.0 mL) at refluxing temperature for 8 h. After concentration and silica gel column chromatography, the nitrile 39 (49.2 mg, 0.266 mmol) and 2-benzylcyclohexanone oxime acetate 41 (8.8 mg, 0.036 mmol) were obtained. in the same manner, the acetate 38 (95.2 mg, 0.30 mmoIl gave the nitriie 40 (45.7 mg, 0.247 mmol) and 41 ( 7.4 mg, 0.030 mmol).

The acetate 37 (63.5 mg, 0.20 mmol) was treated with TMSOTf  $(0.02 \text{ mL})$  in CH<sub>2</sub>Cl<sub>2</sub> (0.6 mL) at O°C for 3 h. After work-up, the nitrile 39 (36.3 mg, 0.18 mmol) was obtained. In the same manner, the acetate 38 (63.5 mg, 0.20 mmoi) gave the nitrile 40 (33.3 mg, 0.19 mmol).

39: a colorless oil; <sup>1</sup>H NMR (CDC1<sub>3</sub>, 60 MHz) 1.33-1.95 (m, 4 H), 1.95-2.78 (m, 4 H), 5.80-6.53 (broad, 2 H, HC=CH), 7.21 (broad s, 5 H); IR (film) 2260, 1601, 1498, 767, 700 cm<sup>-1</sup>. 40: a colorless oil; <sup>1</sup>H NMR (CDCi3, 60 **MHz)** 1.05-1.95 (m, 4 H), 1.95-2.78 (m, 4 H), 5.56 (m, 1 H), 6.40 (m, 1 HI, 7.22 (broad s, 5 H); IR (film) 2260, 1600, 1498, 962 (trans olefin), 742, 692 cm<sup>-1</sup>.

#### **Synthesis of (4E), (7Z)-Tridecadfenyi** Acetate 46.

To a soIutIon of 0.44 mL (2.2 mmol) of hexamethyldisitane In 3.0 mL of **HMPA** was added a solution of methylilthium (2 mmol) at room temperature. After stirring for 30 min, 8 mL of THF was added. The mixture was cooled to -78°C and 123 mg (1.5 mmol) of 2-cyclopentenone was added slowty by microsyrlnge. Then trl-n-butyltin chloride (650 mg, 2.0 mmol) was added. The mixture was stirred for 10 min. 1-Bromo-2-octyne (370 mg, 2.0 mmol) was added. The temperature was raised to  $-40$  °C and the mixture **was** stirred for ih, The mixture was extracted with a mixture of ether and hexane (1:i) and the extract was washed with water and was concentrated. The residual oil was purified by silica gel column to give the silylketone 42 (264 mg, 1.0 mmol). 42: a colorless oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 90 MHz) 0.03 (s, 9 H), 0.84 (t, 3 H), 1.10-1.50 (m, 9 H), 1.70-2.20 (m, 5 HI, 2.30 (m, 1 H), 2.53 (m, 1 H), 2.80 (m, 1 H); IR (film) 1740, 1250, 835 cm<sup>-1</sup>; MS <u>m/e</u> 265 (23, M+1), 264 (6, M), 207 (15), 183 (27), 173 (64), 164 (84), 156 (100).

Under atmospheric pressure of hydrogen, a solution of the siiylketone 42 (136 mg, 0.51 mmoi), Pd-BaSO<sub>4</sub> (30 mg), and quinoline (0.01 mL) in 1.5 mL of methanol was stirred for 10 h. After concentration the residual oil was purified by silica gel column to give the silylketone 43 (1 I8 mg. 0.44 mmol). 43: IR (film) 1738, 1250 cm<sup>-1</sup>; MS m/e 267 (4, M+1), 177 (12), 156 (21), 155 (30), 119 (11), 73 (100).

The silylketone 43 (118 mg, 0.44 mmol) was treated with  $HONH_2$ <sup>+</sup>HCl (37 mg) and NaOAc (72 mg) in ethanol (2.0 mL) to give the corresponding oxime. The crude oxime was treated with acetic anhydride (0.05 mL) and pyridine (0.11 mL) in dichloromethane (1.0 mL) at 0°C for ih. After usual work-up and silica gel column chromatography, the silylketoxime acetate  $44$  (105 mg, 0.32 mmol) and the  $(Z)$ -isomer  $(35 \text{ mg}, 0.08 \text{ mmol})$  were obtained; Rf, 0.50 for 44 and 0.39 for the isomer, ether:hexane = 1:1. 44: a

coloriess oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 90 MHz) -0.06 (s, 9 H), 0.82 (t, 3 H), 2.05 (s, 3 H), 5.20-5.60 (m, 2 H); <sup>13</sup>C NMR (CDC1<sub>3</sub>, 22.5 MHz) -2.75, 13.9, 14.4, 19.6, 21.6, 22.5, 24.8, 27.3, 27.5, 29.3 (two carbons), 30.1, 30.5, 31.5, 44.7, 125.7, 132.2, 168.8, 177.5; IR (film) 1765, 1650, 1245, 835 cm<sup>-1</sup>; MS m/e 264 (10, M-59), 190 (10), 170 (100), 154 (50), 113 (38), 73 (62); MS exact mass  $m/e$  264.2147 (M-AcO) (calcd for C<sub>18</sub>H<sub>32</sub>  $NO<sub>2</sub>Si-C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>$ , 264.2147).

The silylketoxime acetate 44 (109 mg, 0.34 mmol) was treated with TMSOTf (0.07 mL) in CH<sub>2</sub>Cl<sub>2</sub> (1.0 mL) at 0°C for 1 h to give the nitrile 45 (51 mg, 0.27 mmol). 45: a colorless oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 90 MHz) 0.89 (t, 3 H), 1.10-1.50 (m, 6 H), 1.80-2.10 (m, 4 H), 2.34 (s, 1 H), 2.36 (s, 1 H), 2.75 (broad, 2 H), 5.20-5.80 (m, 4 H);  $^{13}$ C NMR (CDCI<sub>3</sub>, 22.5 MHz) 14.0, 17.7, 22.6, 27.2, 28.4, 29.3, 30.2, 31.5, 119.3, 125.9, 126.6, 131.2, 132.2; IR (film) 2240, 965 cm<sup>-1</sup>; MS m/e 191 (16, M), 163 (13), 121 (23), 111 (26), 108 (43), 80 (86), 68 (100); MS exact mass  $m/e$  191.1647 (calcd for  $C_{13}H_{21}N$ , 191.1672).

To a solution of 45 (50 mg, 0.34 mmol) in hexane (1.0 mL) was added di-iso-butylaluminum hydride (0.08 mL) by micro-syringe at -78°C. After stirring for 30 min, ether(2 mL) was added and then hydrochloric acid was added until the pH of the mixture became ca. 4. After separation of the organic layer and subsequent concentration, the residual oil was diluted with methanol  $(1 \text{ mL})$  at  $0^{\circ}$ C. Suspension of sodium borohydride (20 mg) in methanol (0.5 mL) was added. The mixture was stirred at  $0^{\circ}C$  for 1h. After an extraction, the residual oil was obtained. The oil was treated with acetic anhydride (0.04 mL) and pyridine (0.06 mL) in CH<sub>2</sub>Cl<sub>2</sub> (1.0 mL) at 0°C for 2 h. After usual work-up and silica gel column chromatography, the desired acetate 46 (36 mg, 0.18 mmol) was obtained. 46: a colorless oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 90 MHz) 0.89 (t, 3 H), 2.04 (s, 3 H), 2.74 (s, 2 H), 4.06 (t, 2 H), 5.20-5.60 (m, 4 H); <sup>13</sup>C NMR  $(CDC1_{3}$ , 22.5 MHz) 14.0, 20.9, 22.6, 27.1, 28.5, 28.9, 29.4, 30.4, 31.5, 64.0, 127.4, 129.1, 129.6, 130.7, 172.0; IR (film) 1740, 965 cm<sup>-1</sup>; MS exact mass m/e 238.1926 (calcd for C<sub>15</sub>H<sub>26</sub>O<sub>2</sub>, 238.1933); MS m/e 238 (8, M), 179 (21), 150 (29), 135 (70), 122 (51), 108 (58), 97 (35), 78 (42), 57 (48), 32 (100).

## Synthesis of the Ketones 47 and 48.

The nitrile 32 (5.1 mg) was treated with n-decanyl magnesium chloride in ether at 0°C. After treatment with hydrochloric acid, the desired ketone  $47$  (6.7 mg, 0.022 mmol) was obtained. The nitrile 36 (9.3 mg) was also treated by the same manner to give the nitrile  $48$  (13.3 mg, 0.043 mmol); Rf, 0.72 for 47 and 0.62 for 48, ether:hexane = 1.4. 47: mp 34.0-35.0 °C; <sup>1</sup>H NMR (CDC1<sub>3</sub>, 90 MHz) 0.87 (t, 3 H) 1.0-1.8 (m, 28 H), 1.80-2.20 (m, 4 H), 2.37 (t, 4 H), 5.25-5.45 (m, 2 H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 22.5 MHz) 129.3 and 131.6 (olefinic); IR (film) 1700, 960 cm<sup>-1</sup>; MS m/e 308 (33, M), 251 (13), 237 (16), 197 (23), 169 (66), 124 (100). 48: a coloriess oil; 'H NMR (CDCl<sub>3</sub>, 90 MHz) 0.88 (t, 3 H), 1.0-1.4 (m, 24 H), 1.60 (m, 4 H), 1.95 (m, 4 H), 2.32 (t, 4 H), 5.10-5.60 (m, 2 H); <sup>16</sup>C NMR (CDC1<sub>3</sub>, 22.5 MHz) 128.8 and 131.3 (olefinic); IR (film) 1710 cm<sup>-</sup>; MS <u>m/e</u> 308 (21, M), 197 (22), 167 (22), 124 (90).

Acknowledgment. We gratefully acknowledge the support of the Ministry of Education, Science and Culture by Grant-in-Aid for Scientific Research (61470087) and by Grant-in-Aid for Special Project Research (61125004).

## References and Notes.

- (1) Maruoka, K.; Miyazaki, T.; Ando, M.; Matsumura, Y.; Sakane, S.; Hattori, K.; Yamamoto, H. J. Am. Chem. Soc. 1983, 105, 2831; for reviews, references cited therein.
- (2) (a) Donarum, L. G.; Heldt, W. Z. Org. React.; John Wiley & Sons, Inc.: New York, 1960;  $11$ , p 1-156. (bl McCarty, C. G. In " The Chemistry of the Carbon-Nitrogen Double Bond: Patai, S., Ed.; Wiiey-Interscience: New York, 1970; p 416-439. (c) Grob, C. A. Angew. Chem., Int. Ed. Engl. 1969, 8, 535-546. (d) Marshall, J. A.; Anderson, N. H.; Johnson, P. C. J. Org. Chem. 1970, 35, 186. (e) Kocienski, P. J.; Ansell, J. M. Ibid, 1977, 42, 1102. (f) Wenkert, E.; Berges, D. A.; Golob, N. F. J. Am. Chem. Soc. 1978, 100, 1263. (g) Ayer, W. A.; Browne, L. M. Can. J. Chem. 1974, 96, 5270.
- (3) Fleming, I. Chem. Soc. Rev. 1981, 10, 83.
- (41 (al Hudrlik, P. F.; Hudrlik, A. M.; Nagendrappa, G.; Yimenu, T.; Zeliers, E T.; Chin, E J. Am. Chem. Soc. 1980, 102, 6896. (b) Denmark, S. E.; Jones, T. K. Ibid, 1982 104, 2624. (c) Bac, N. V.;

Langlois, Y. *Ibid*, 1982, 104, 7666.

- (5) **Our** preliminary communications on this work have appeared: (a) Nishiyama, H.; Sakuta, K.; Osaka, N.; Itoh, K. Tetrahedron Lett. 1983, 24, 4021. (b) Nishiyama, H.; Sakuta, K.; Itoh, K. Ibid, 1984, **25\_** 223.
- (6) See ref. 2(b).
- **(7)**  Hudrlik, P. F.; Waugh, M. A.; Hudrlik, A. M. J. Organomet. Chem. **1984, 271, 69.**
- (8) Ine lactams 5 and 9 were major products from 7 and 8 by use of  $\text{PCl}_5$ ,  $\text{TiCl}_4$ , or AICl<sub>3</sub>, respective-JY.
- **(9)**  (a) Shiono, M.; Echigo, Y.; Mukalyama, T. Chem. Lett. **1976,** 1397. (b) Mukaiyama, T. Angew. Chem. Int. Ed. Engl. 1979, 18, 707.
- **(10)**  From 19, 3-trimethylsilylpropionitrlle was also obtained in ca. 20% yield by way of fragmentation of the t-butyl group.
- (11) In the course of the reactlon, the corresponding (El-isomer could not be detected by careful TLC examination. The (E)-isomer generated by the isomerlzation should spontaneously cause the fragmentation under the reaction condition.
- (12) (a) Noyori, R.; Murata, S.; Suzuki, M. Tetrahedron **1981, 3J, 3899.** (b) Emde, H.; Domsch, D.; Feger, H.; Frick, U.; Gotz, A.; Hergott, H. H.; Hofmann, K.; Kober, W.; Krageloh, K.; Oesterle, T.; Steppan, W.; West, W.; Simchen, G. Synthesis **1982,** 1. (c) Ishida, Y.; Sasatani, S.; Maruoka, K.; Yamamoto, H. Tetrahedron Lett. 1983, 24, 3255. (d) Sakane, S.; Matsumura, Y.; Yamanura, Y.; Ishida, Y.; Maruoka, K.; Yamamoto, H. J.Am. Chem. Soc. 1983, 105, 672. (e) Matsumura, Y.; Fujiwara, J.; Maruoka, K.; Yamamoto, H. Ibid. **1983. 105,** 6312.
- **(13)**  <sup>1</sup>H NMR (90 MHz): for 37,  $J(H\alpha - H\beta') = 12.5$  Hz; for 38,  $J(H\alpha - H\beta') = 11.5$  Hz.
- **(14)**  (a) Roelofs, W. L.: Kochansky, J. P.; Carde, R. T.; Henrick, C. A.: Labovitz, J. N.; Corbin. V. L. Life Sci. 1975, 11, 699. (b) Alexakis, A.; Cahiez, G.; Norman, J. F. Tetrahedron Lett. **1978,** 2027.
- (15) Still, W. C. <u>J. Org. Chem.</u> 1**976**, <u>41</u>, 3063.
- (16) (a) Odic, Y.; Pereyre, M. <u>J. Organomet. Chem.</u> 1973, 55, 273. (b) Tardella, P. A. <u>Tetrahedron Lett</u> 1969, 1117. (c) Suzuki, M.; Yanagisawa, A.; Noyori, R. J. Am. Chem. Soc. 1985, 107, 3348.
- (17) Smith, R. G.: Daves, G. D. <u>J. Org. Chem.</u> 1976, <u>41</u>, 3063
- **(18)**  Gras, J.-L. Organic Syntheses; Wiely: New York, 1981; Vol. 60, p 88.
- **(19)**  Taber, D. F.; Gunn, B. P.; Chiu, I.-C. Organic Syntheses; Wiley: New York, Vol. 61, p 59.